



UNIVERSITY OF
MARYLAND



Scaling down the laws of thermodynamics

What do the laws of thermodynamics look like
when applied to very small systems?

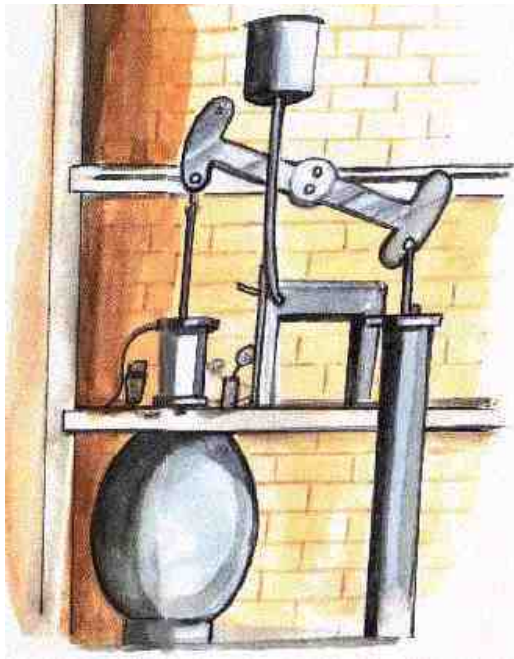
Chris Jarzynski

Institute for Physical Science and Technology
Department of Chemistry & Biochemistry
Department of Physics

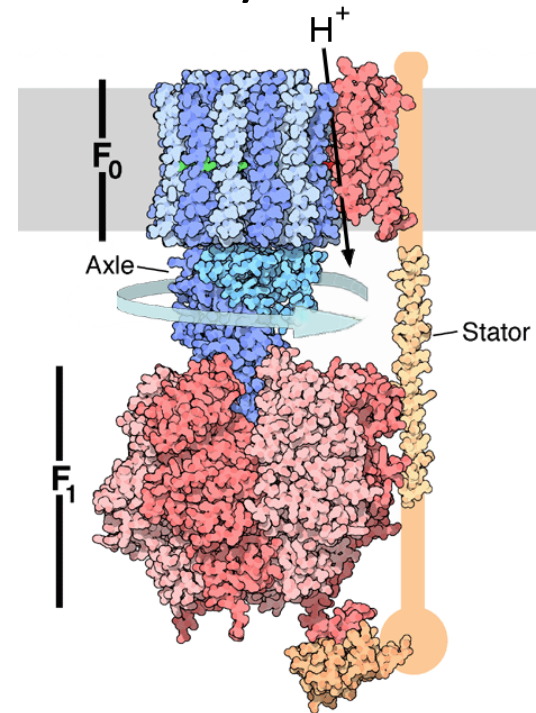


Macroscopic and microscopic machines

steam engine



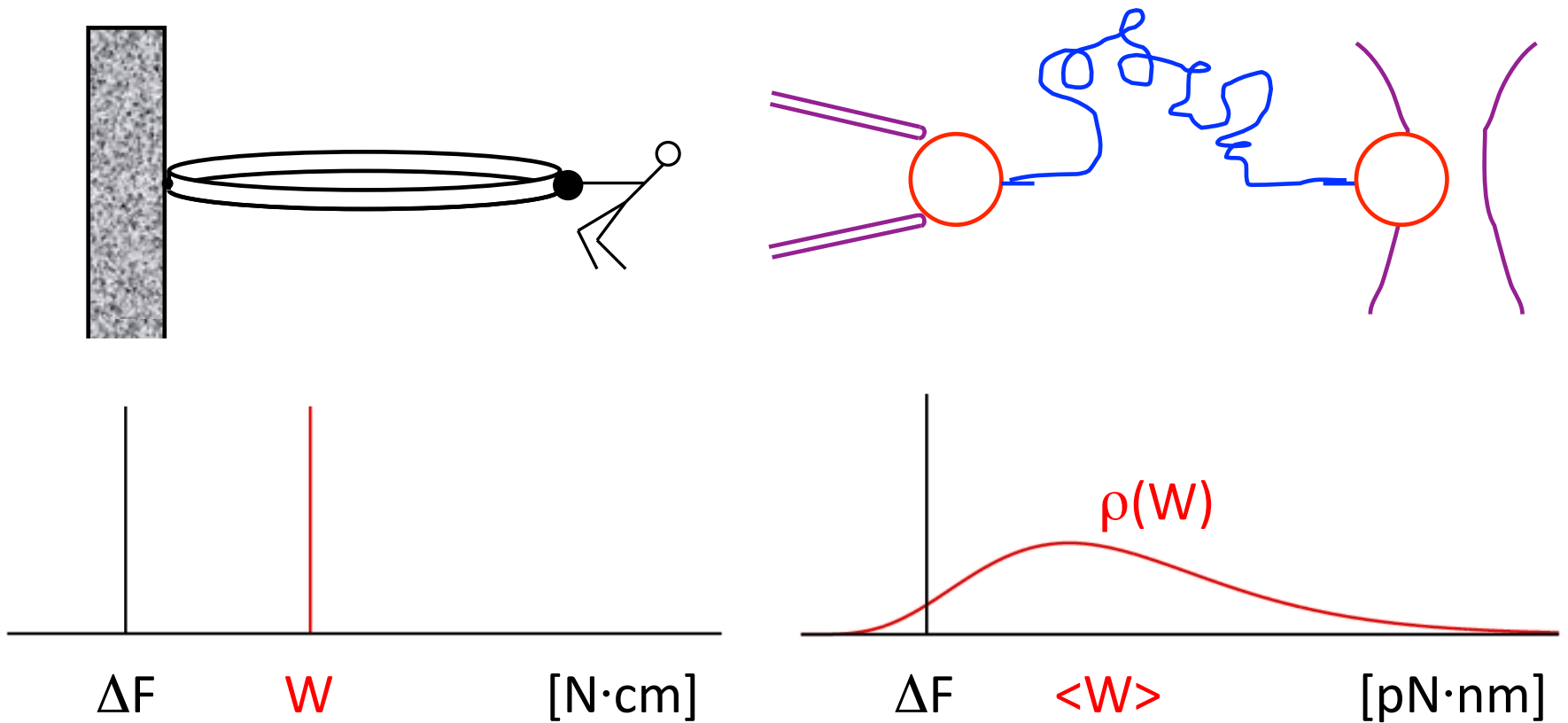
ATP synthase



RCSB Protein
Data Bank

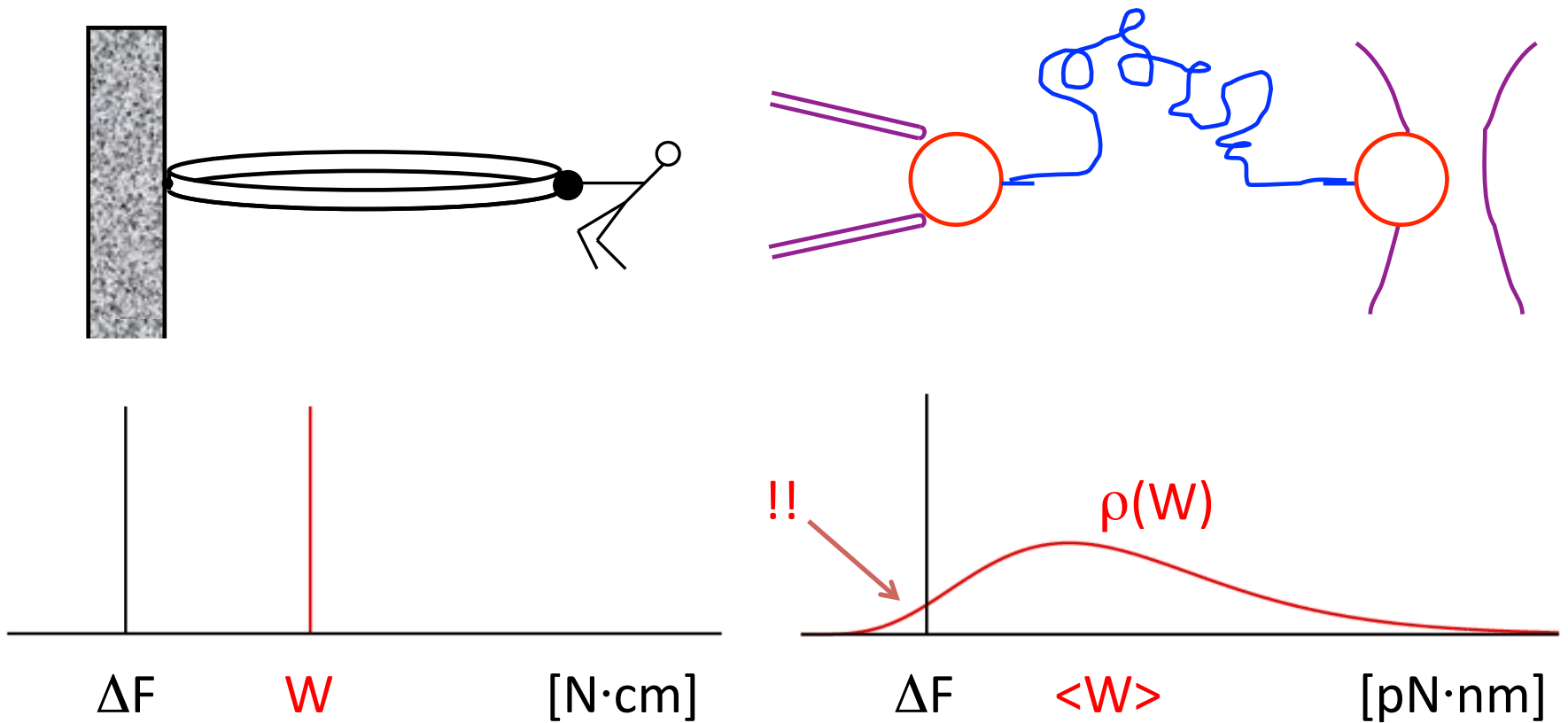
New features of thermodynamics at the nanoscale

- Prominence of fluctuations



New features of thermodynamics at the nanoscale

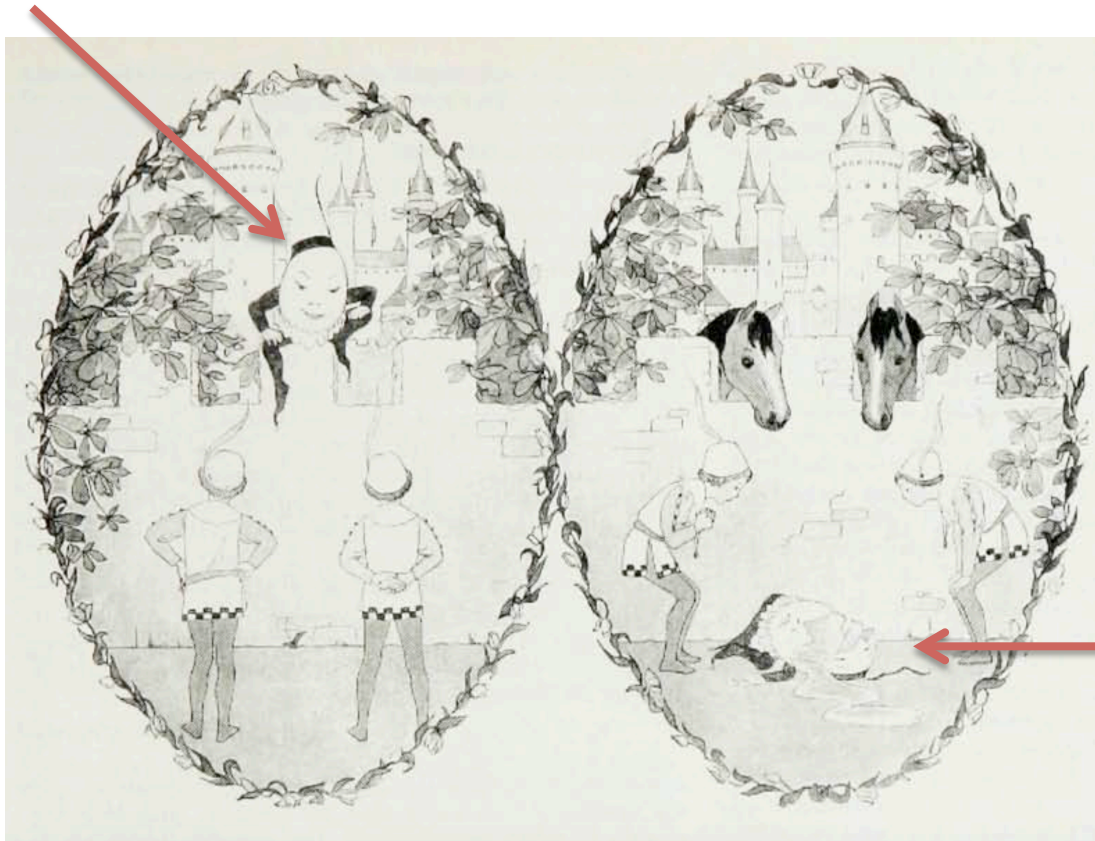
- Prominence of fluctuations
- **“Violations” of the second law**



New features of thermodynamics at the nanoscale

- Prominence of fluctuations
- “Violations” of the second law
- **Blurred arrow of time**

happy
Humpty

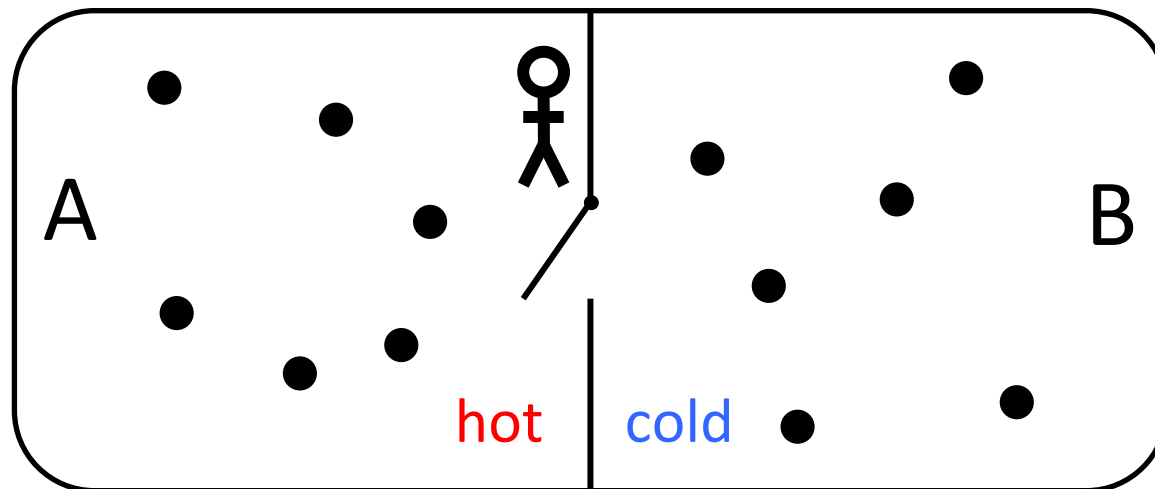


J.L. Lebowitz,
Physics Today (1993)

broken
Humpty

New features of thermodynamics at the nanoscale

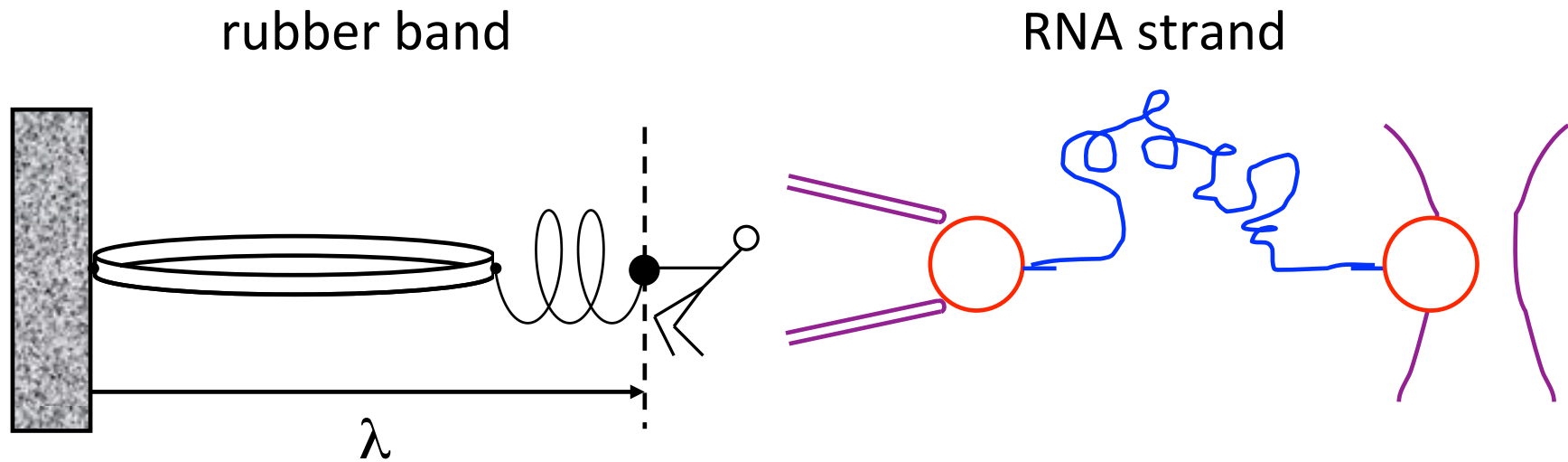
- Prominence of fluctuations
- “Violations” of the second law
- Blurred arrow of time
- **Feedback control & information processing**



New features of thermodynamics at the nanoscale

- Prominence of fluctuations
- “Violations” of the second law
- Blurred arrow of time
- Feedback control & information processing
- **Strong system-environment coupling**

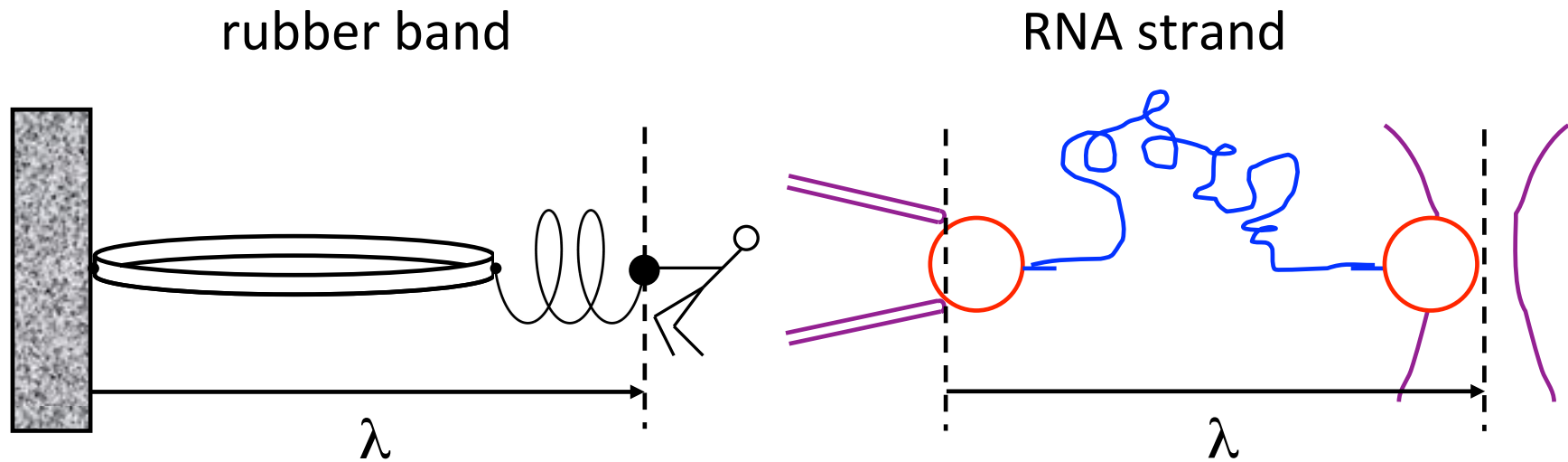
Macro- and nanoscale thermodynamic processes



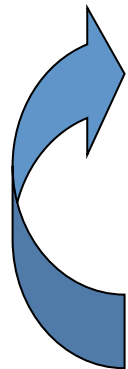
Irreversible process (rubber band):

1. Begin in equilibrium $\lambda = A$
2. Stretch the system $\lambda : A \rightarrow B$
 $W = \text{work performed} \geq \Delta F = F_B - F_A$
3. End in equilibrium $\lambda = B$

Macro- and nanoscale thermodynamic processes



Irreversible process (RNA):



1. Begin in equilibrium
2. Stretch the system
3. End in equilibrium

$$\lambda = A$$

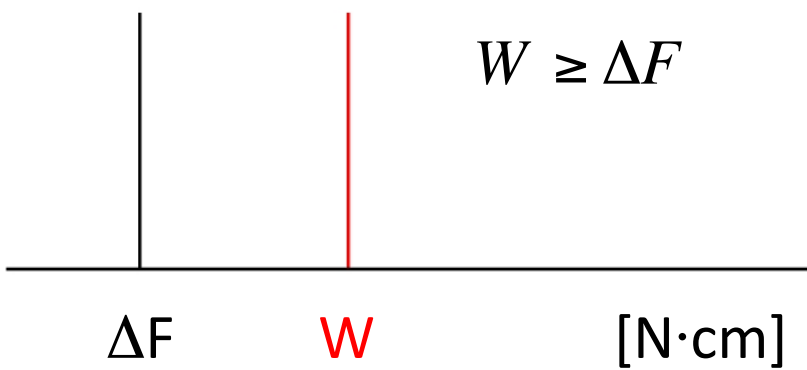
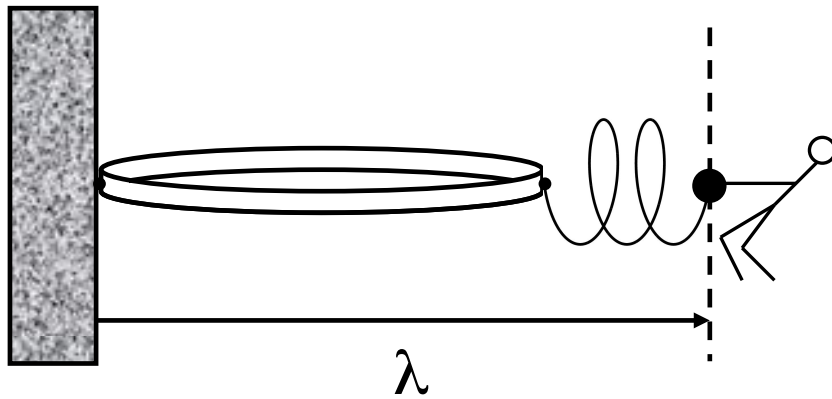
$$\lambda : A \rightarrow B$$

$$\langle W \rangle = \text{average work} \geq \Delta F = F_B - F_A$$

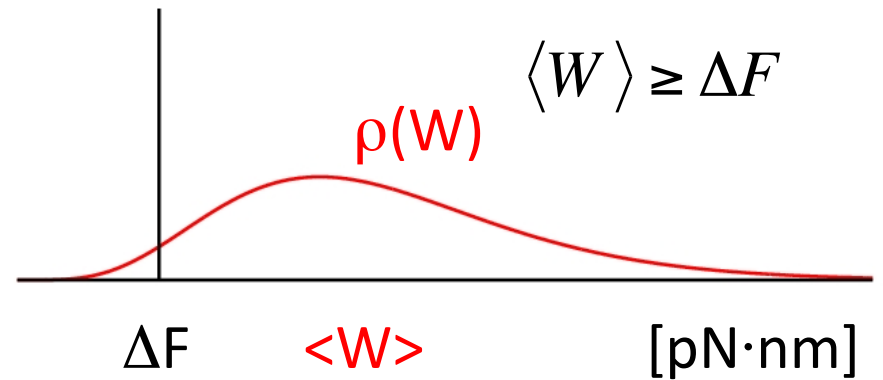
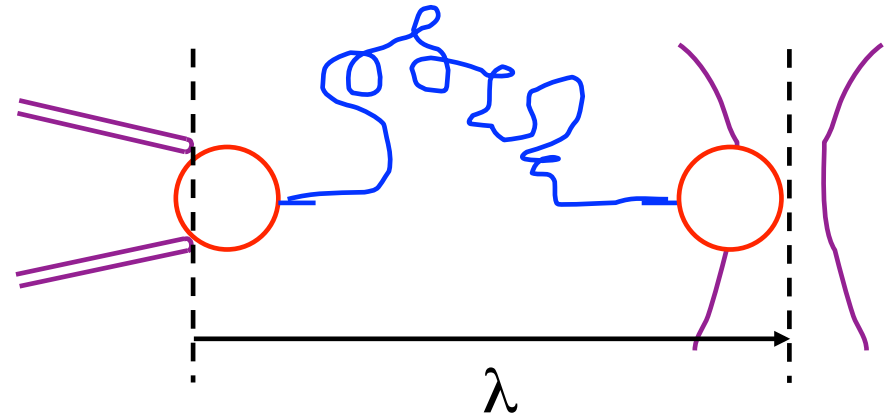
$$\lambda = B$$

Macro- and nanoscale thermodynamic processes

rubber band

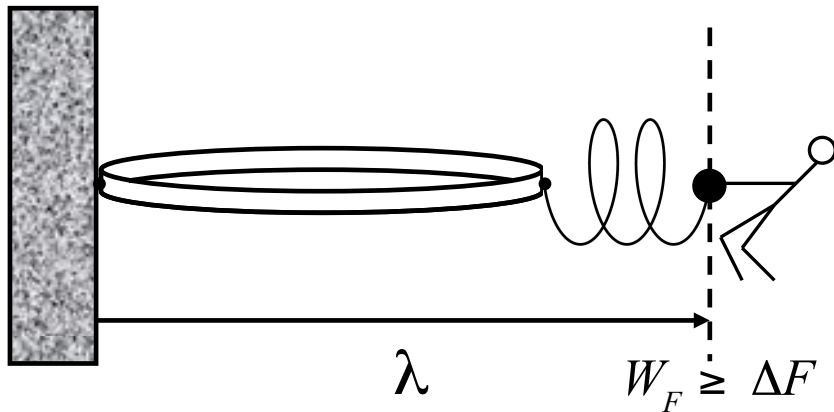


RNA strand

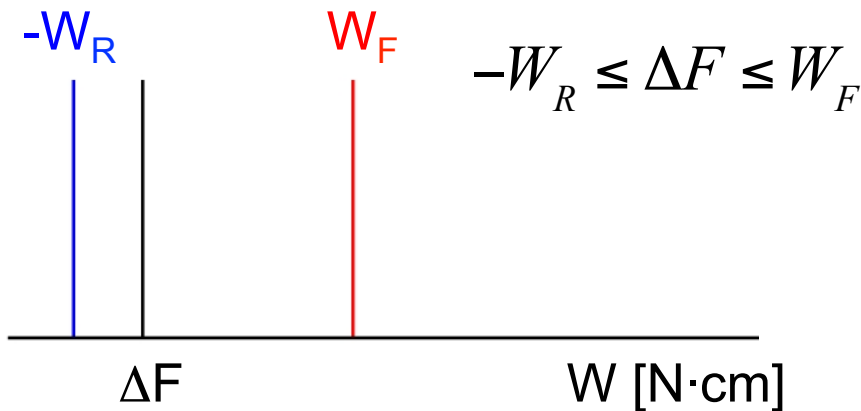


Macro- and nanoscale thermodynamic processes

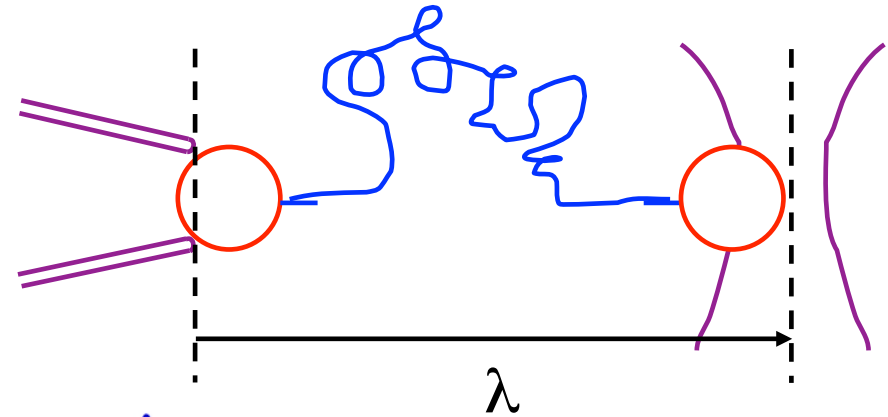
rubber band



$$W_R \geq -\Delta F$$



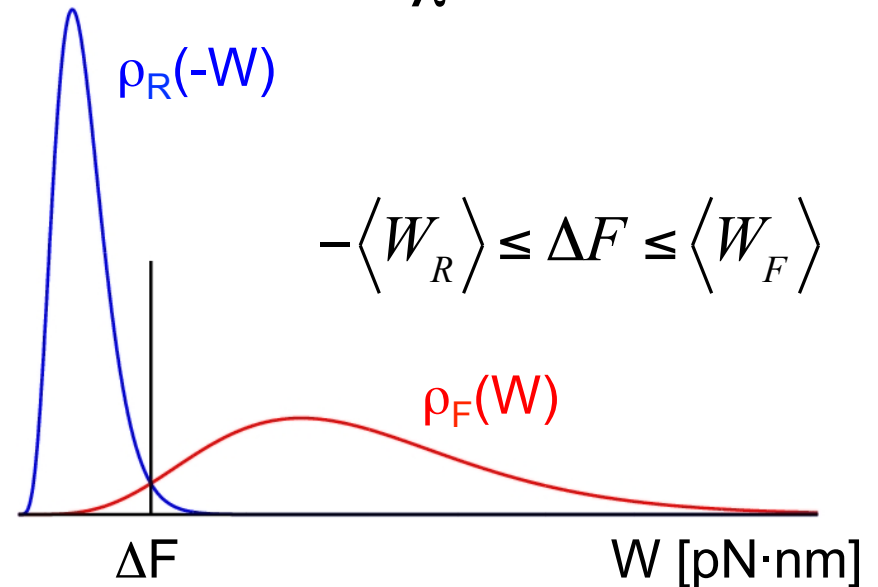
RNA strand



$\rho_R(-W)$

$$-\langle W_R \rangle \leq \Delta F \leq \langle W_F \rangle$$

$\rho_F(W)$

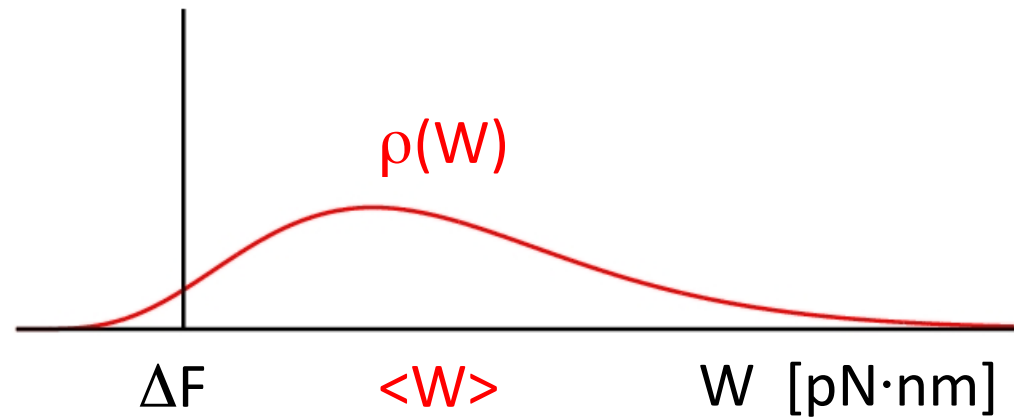


Fluctuations in W satisfy unexpected laws.

Fluctuation theorems / non-equilibrium work relations

$$\left\langle e^{-\beta W} \right\rangle = e^{-\beta \Delta F}$$

C.J., *PRL* **78**, 2690 (1997)



Fluctuations in W satisfy unexpected laws.

Fluctuation theorems / non-equilibrium work relations

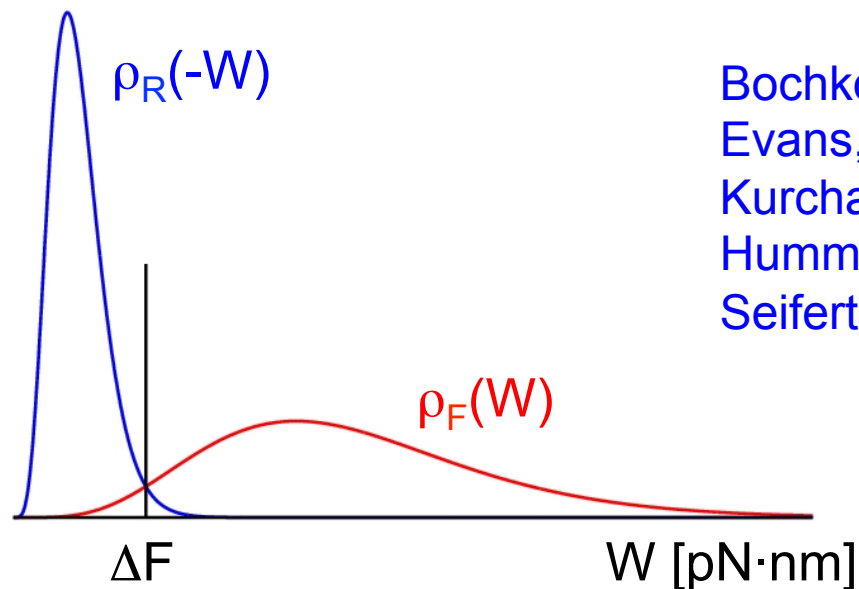
$$\left\langle e^{-\beta W} \right\rangle = e^{-\beta \Delta F}$$

C.J., *PRL* **78**, 2690 (1997)

$$\frac{\rho_F(+W)}{\rho_R(-W)} = \exp[\beta(W - \Delta F)]$$

Crooks, *PRE* **60**, 2721 (1999)

[*J Stat Phys* **90**, 1481 (1998)]



Bochkov & Kuzovlev

Evans, Cohen, Morriss, Searles, Gallavotti

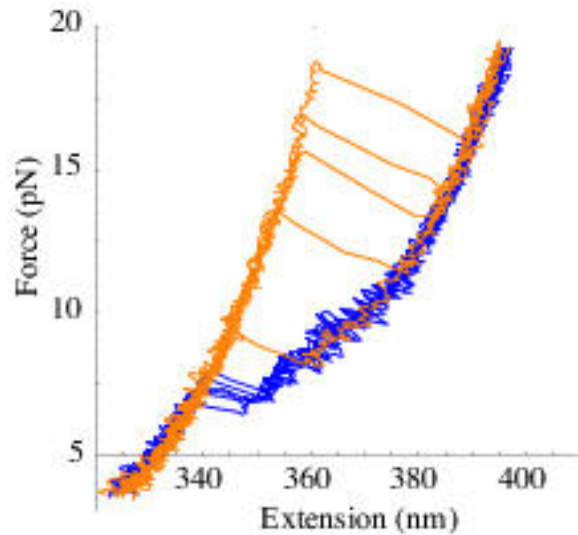
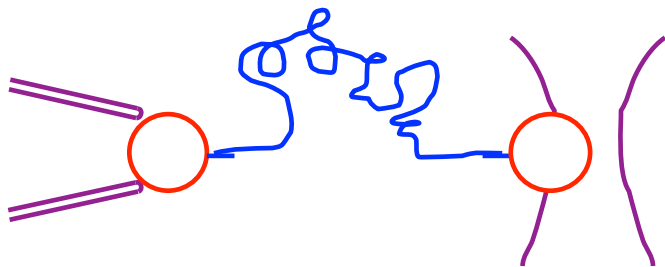
Kurchan, Lebowitz, Spohn

Hummer & Szabo

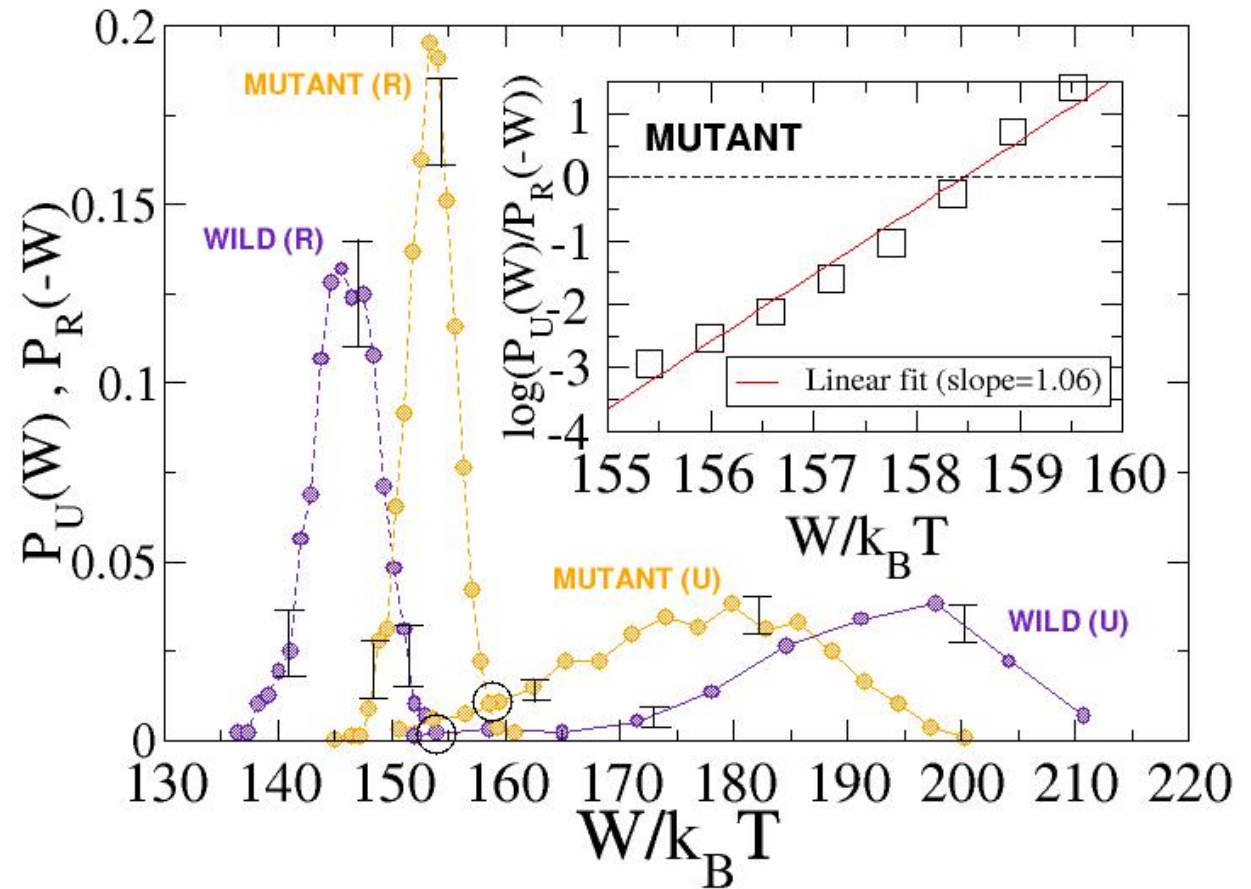
Seifert ...

Unfolding & refolding of ribosomal RNA

$$\frac{\rho_{unfold}(+W)}{\rho_{refold}(-W)} = \exp[\beta(W - \Delta F)]$$



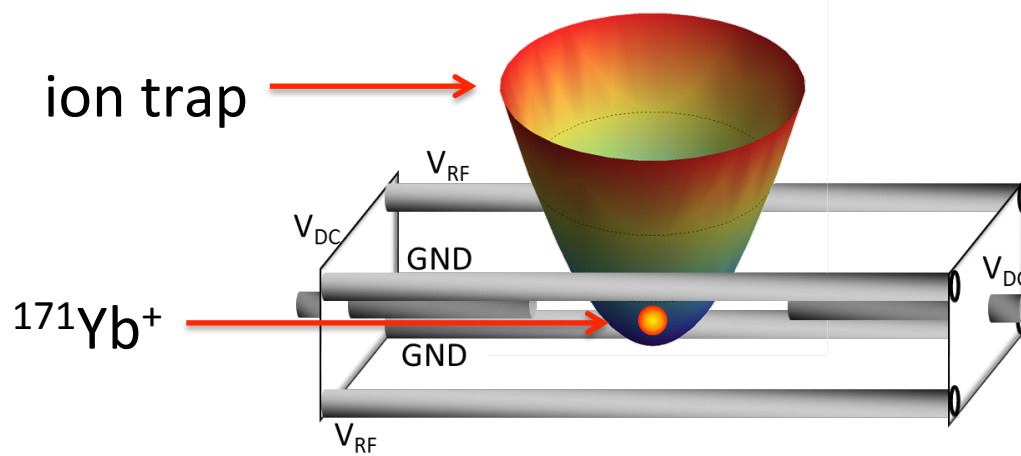
Collin *et al*, *Nature* **437**, 231 (2005)



Quantum nonequilibrium work relation $\langle e^{-\beta W} \rangle = e^{-\beta \Delta F}$

Mukamel, *PRL* **90**, 170604 (2003)

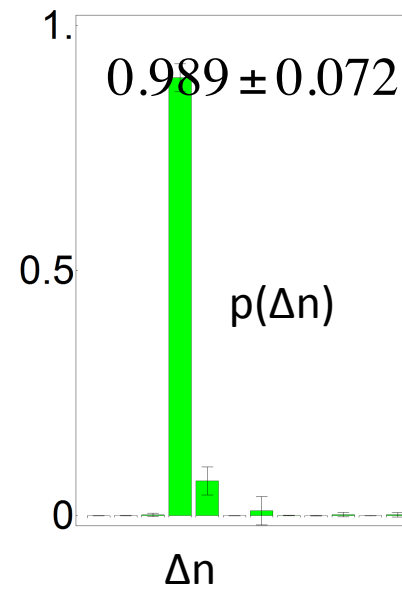
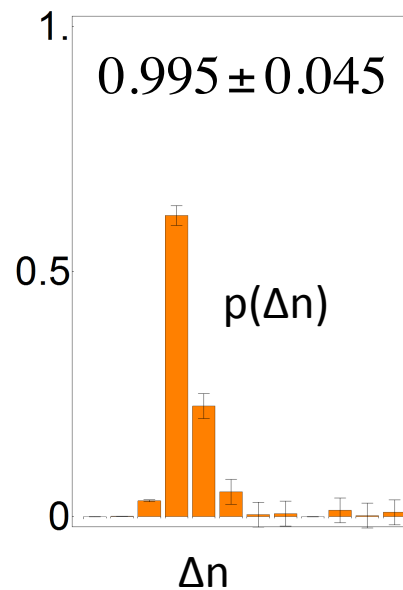
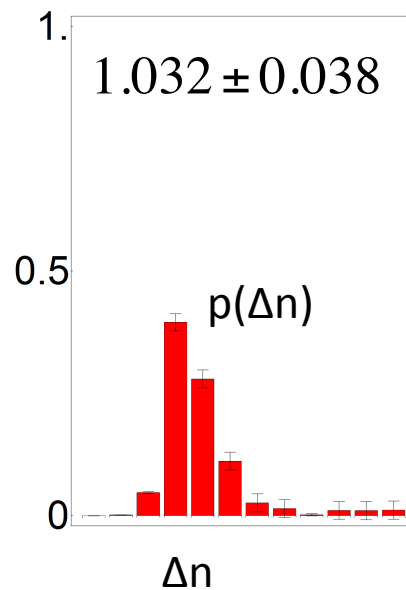
Kurchan, cond-mat/0007360 ; Tasaki, cond-mat/0009244



$$E_n = \hbar\omega \left(n + \frac{1}{2} \right)$$

$$W = \hbar\omega (n_f - n_i) = \hbar\omega \Delta n$$

An et al,
Nat. Phys. **11**, 193 (2015)



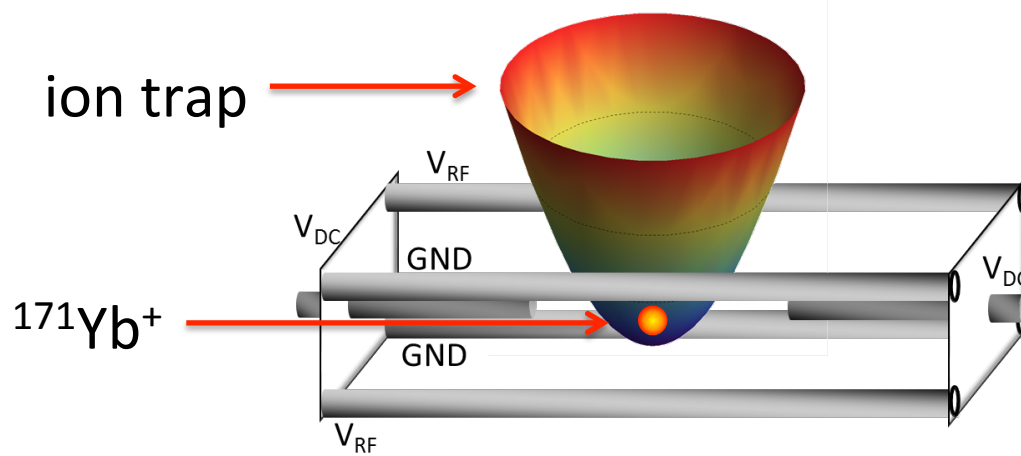
$$\Delta F = 0$$

$$\langle e^{-\beta W} \rangle = 1$$

Quantum nonequilibrium work relation $\langle e^{-\beta W} \rangle = e^{-\beta \Delta F}$

Mukamel, *PRL* **90**, 170604 (2003)

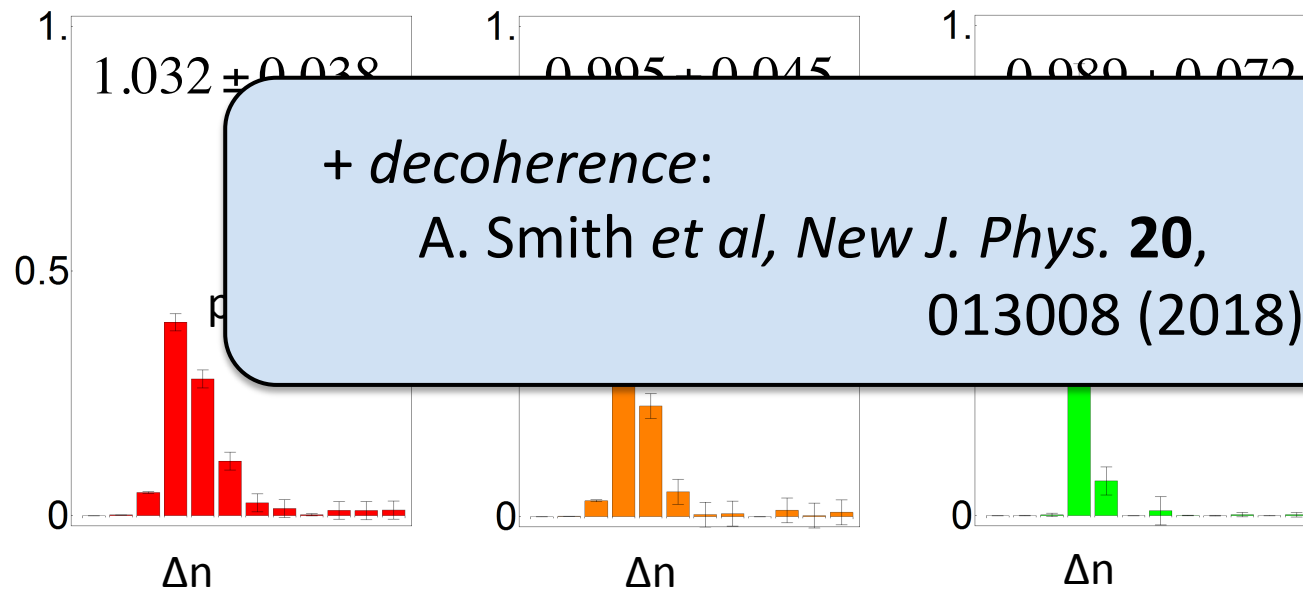
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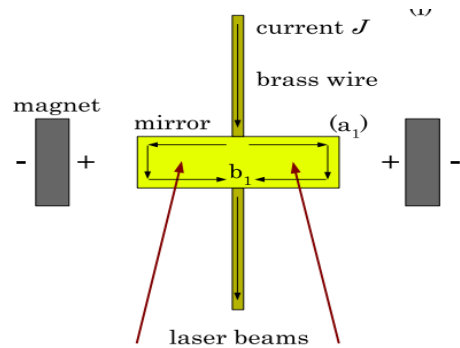
An et al,
Nat. Phys. **11**, 193 (2015)



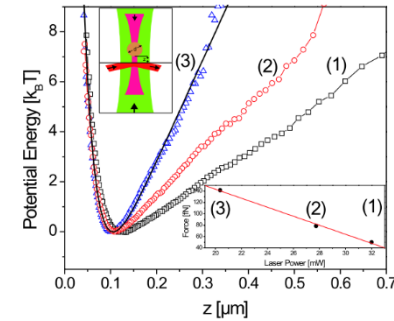
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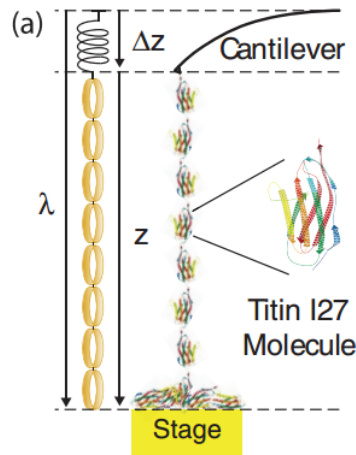
Further experimental verification



Mechanical oscillator
 Douarche *et al*, *EPL* **70**, 593 (2005)

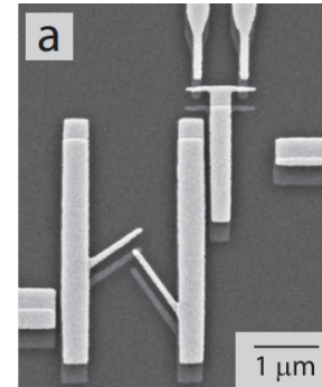


Trapped colloidal particle
 Blickle *et al*, *PRL* **96**, 070603 (2006)



$$\langle e^{-\beta W} \rangle = e^{-\beta \Delta F}$$

$$\frac{\rho_{unfold}(+W)}{\rho_{refold}(-W)} = \exp[\beta(W - \Delta F)]$$



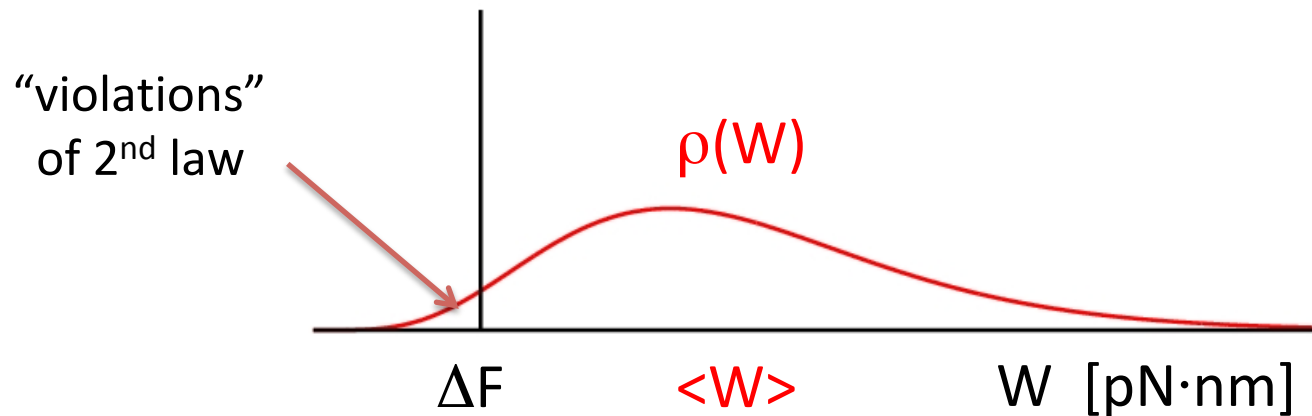
Protein unfolding
 Harris, Song and Kiang,
PRL **99**, 068101 (2007)

Single electron box
 Saira *et al*, *PRL* **109**, 180601 (2012)

& others ...

Implications for the Second Law

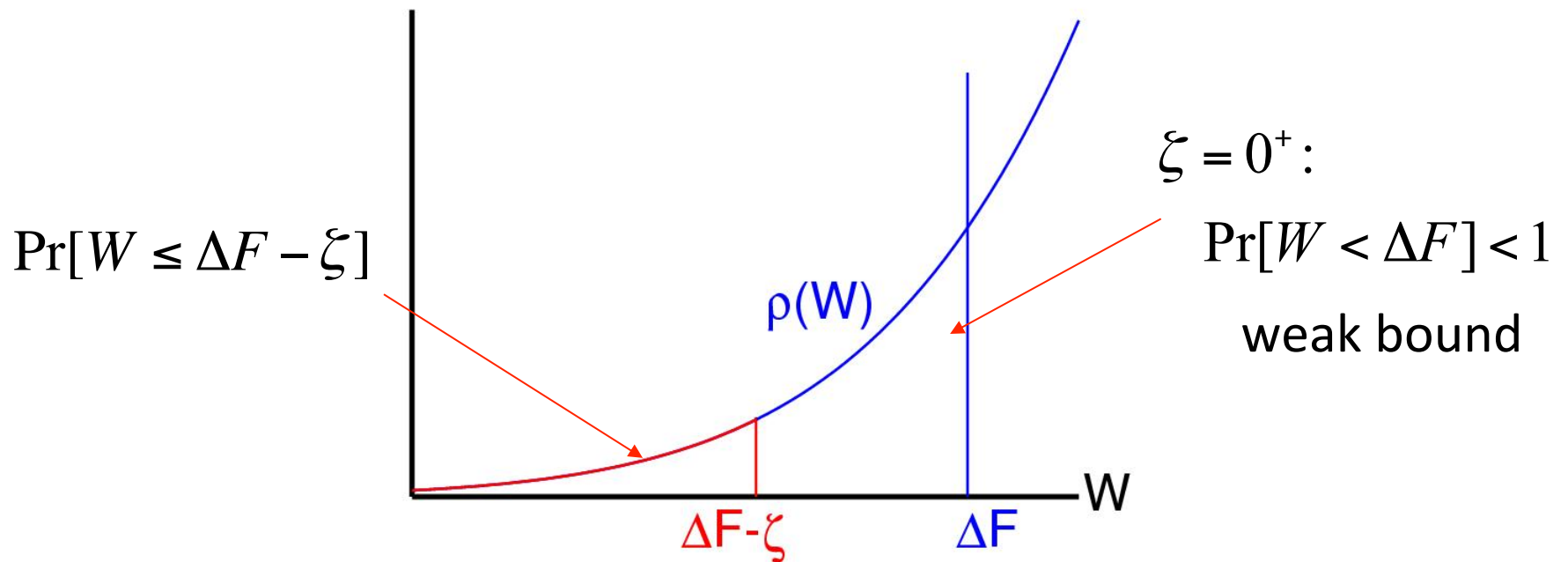
$$\langle e^{-\beta W} \rangle = e^{-\beta \Delta F} \quad \text{implies} \quad \left\{ \begin{array}{l} \langle W \rangle \geq \Delta F \\ \Pr[W \leq \Delta F - \zeta] \leq \exp(-\zeta / k_B T) \end{array} \right.$$
$$\langle e^x \rangle \geq e^{\langle x \rangle}$$



Implications for the Second Law

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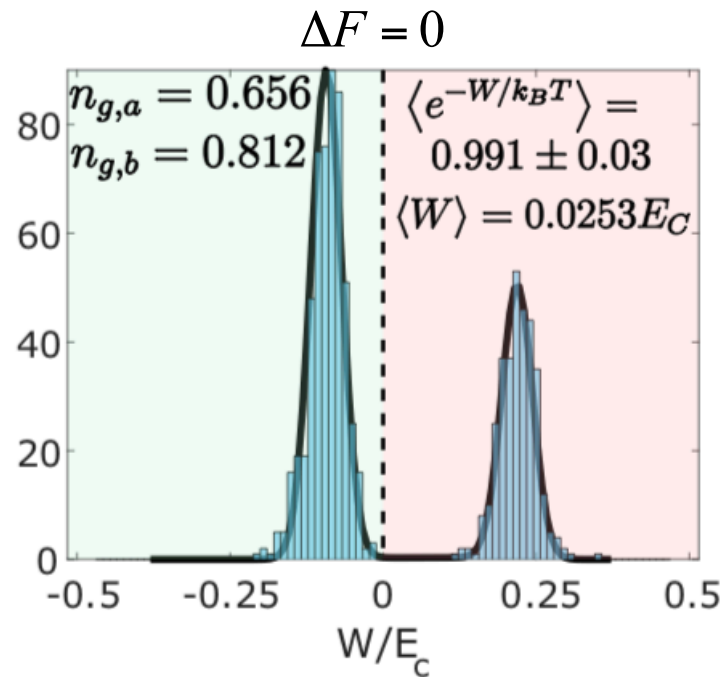
What is the probability that the 2nd law is “violated” by at least ζ ?



Implications for the Second Law

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What is the probability that the 2nd law is “violated” by at least ζ ?



$$\Pr[W < \Delta F] \approx 0.65$$

$$\langle W \rangle > \Delta F$$

Single electron transistor
Maillet *et al*,
arXiv 1810:06274

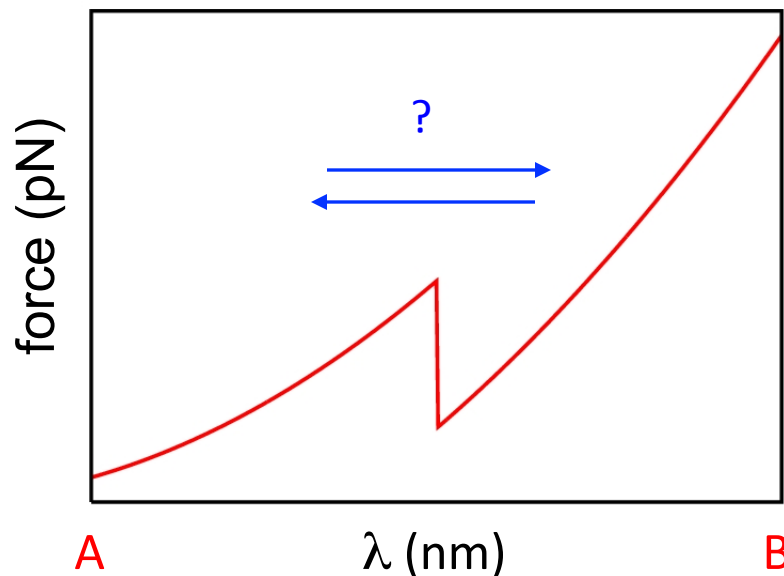
Guessing the direction of the arrow of time

C.J., *Annu Rev Cond Matt Phys* **2**, 329 (2011)

You are shown a movie depicting a thermodynamic process, $A \rightarrow B$.

Task: determine whether you are viewing the events in the order in which they actually occurred, or a movie run backward of the reverse process.

e.g.

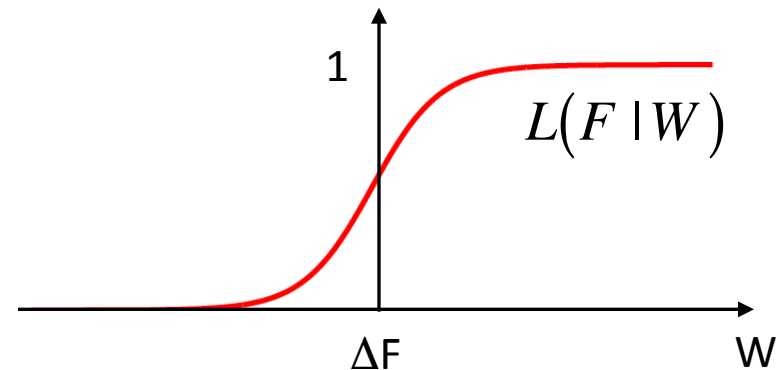


Two hypotheses:

The molecule was stretched (F)

The molecule was contracted (R)

$$L(F | W) = \frac{1}{1 + \exp[-\beta(W - \Delta F)]}$$



Shirts *et al*, *PRL* **91**, 140601 (2003),
Maragakis *et al*, *JCP* **129**, 024102 (2008)

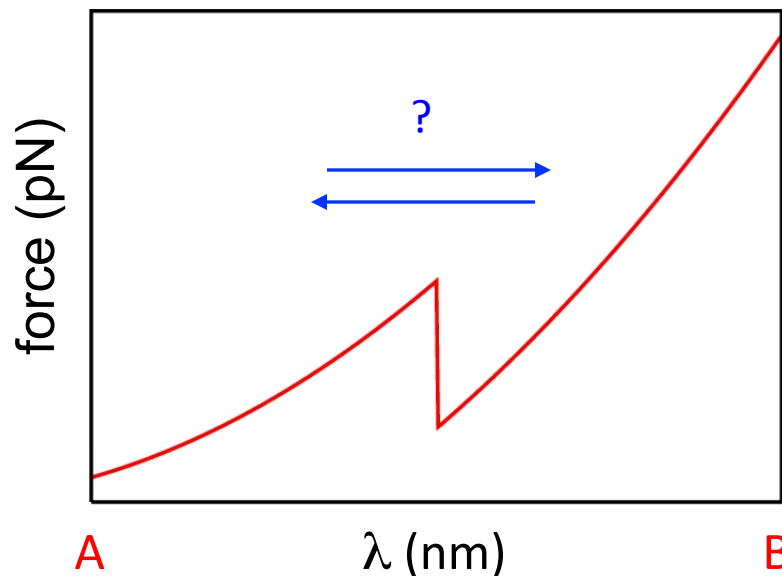
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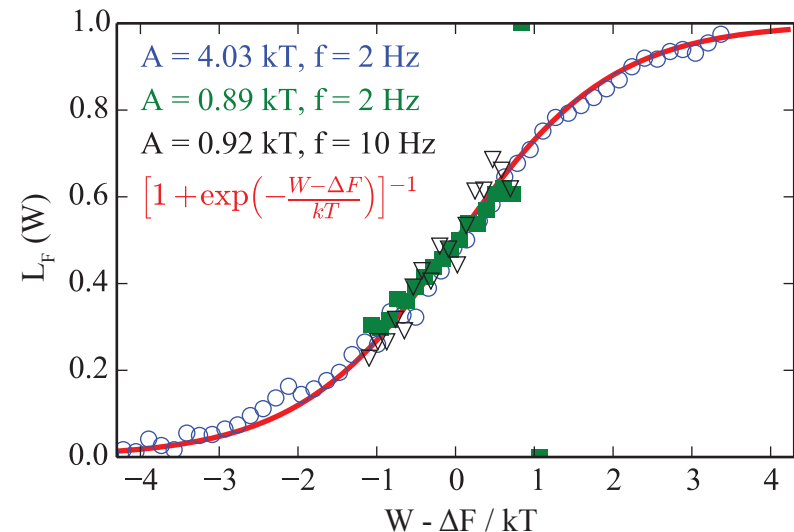
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Hofmann *et al*, *Phys Status Solidi* **254**, 1600546 (2017)

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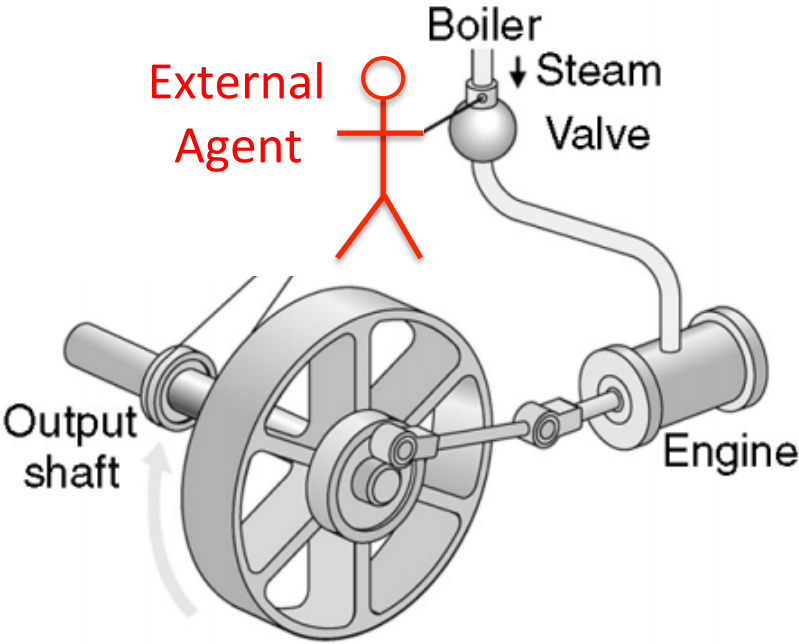
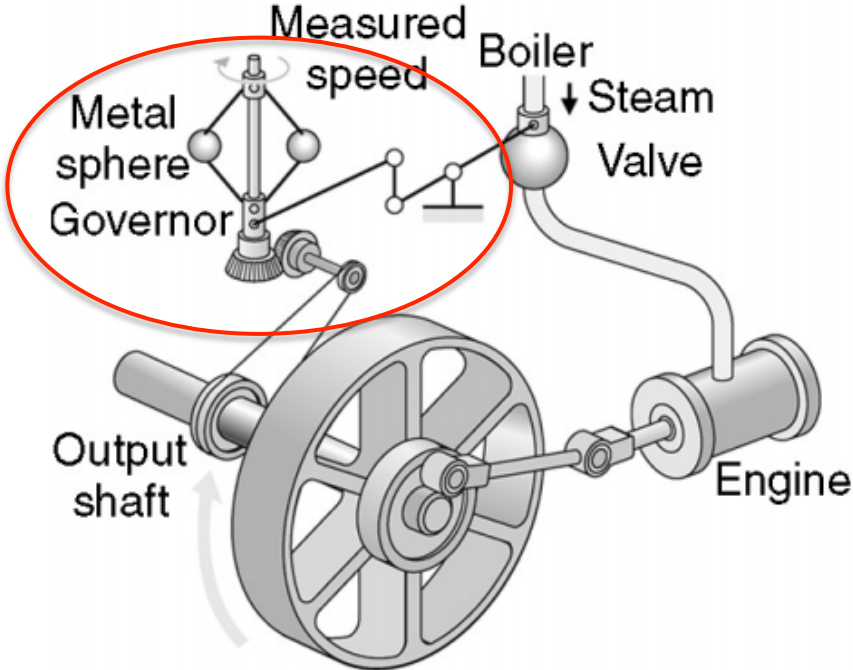
Shirts *et al*, *PRL* **91**, 140601 (2003),
Maragakis *et al*, *JCP* **129**, 024102 (2008)



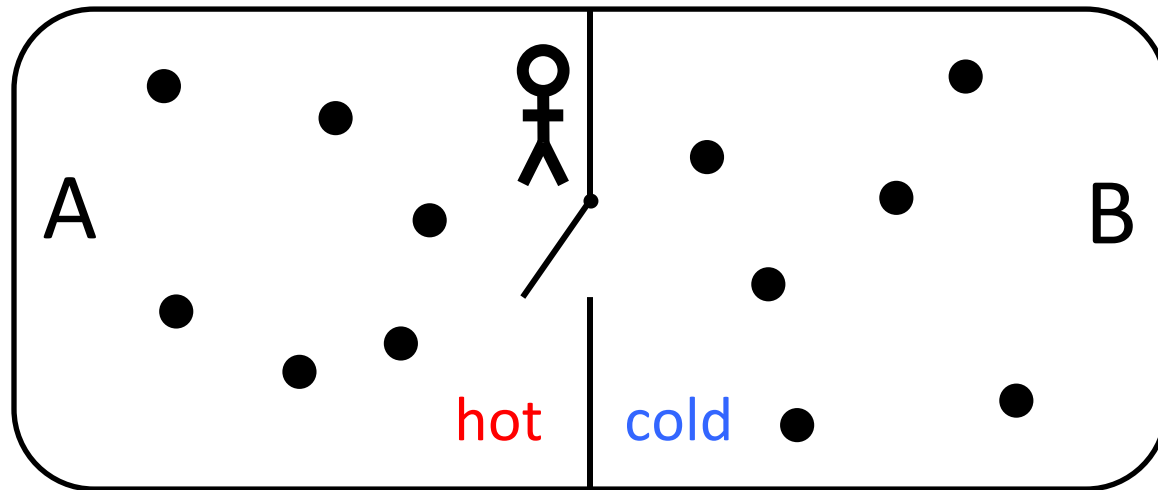
Feedback control

autonomous

non-autonomous



Maxwell's demon

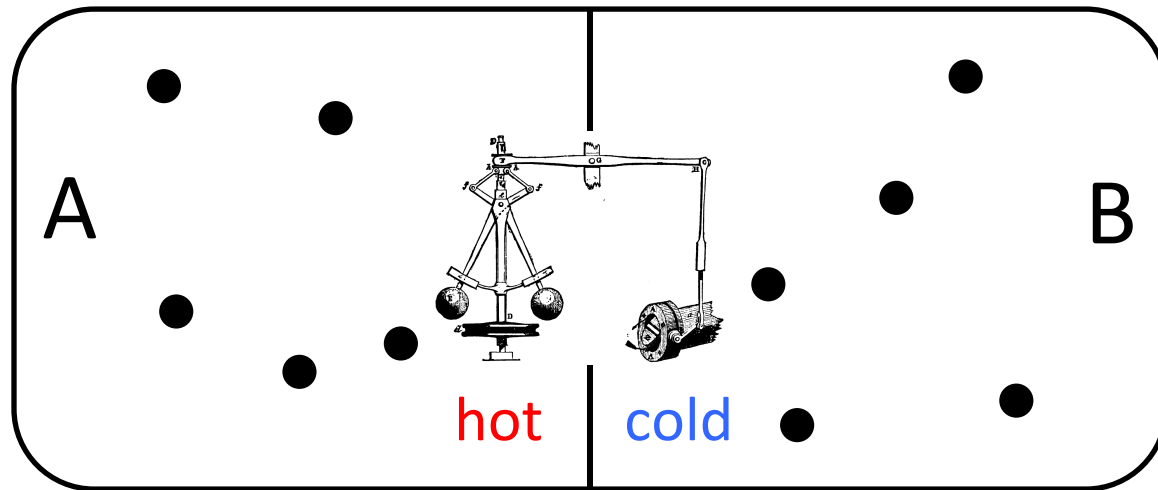


“... the energy in A is increased and that in B diminished; that is, the hot system has got hotter and the cold colder and yet no work has been done, only the intelligence of a very observant and neat-fingered being has been employed”

J.C. Maxwell, letter to P.G. Tait, Dec. 11, 1867

non-autonomous feedback control

Maxwell's demon



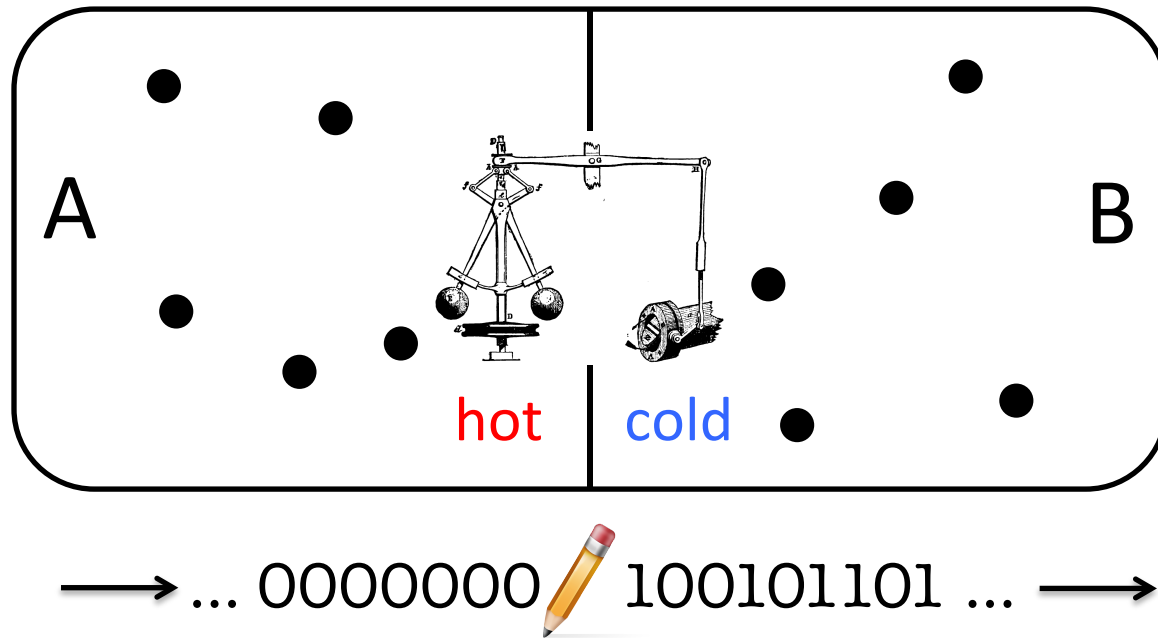
Is a “mechanical” Maxwell demon possible?

M. Smoluchowski, *Phys Z* **13**, 1069 (1912) **no!**

R.P. Feynman, *Lectures*

autonomous feedback control

Maxwell's demon



Is a "mechanical" Maxwell demon possible?

R. Landauer, *IBM J Res Dev* **5**, 183 (1961)

O. Penrose, *Foundations of Statistical Mechanics* (1970)

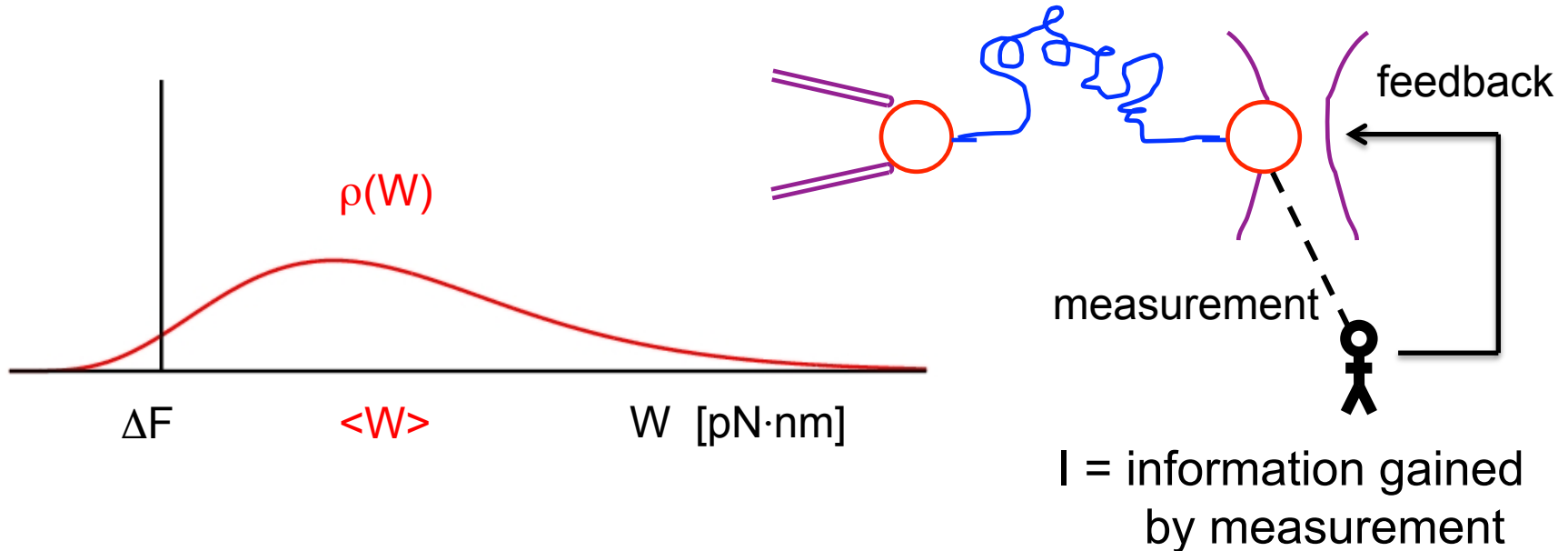
C.H. Bennett, *Int J Theor Physics* **21**, 905 (1982)

yes, but ...

autonomous feedback control

Second Law of Thermodynamics

... with measurement and feedback



$$\langle W \rangle \geq \Delta F - k_B T \langle I \rangle$$

Sagawa & Ueda, *PRL* **100**, 080403 (2008)

$$\langle e^{-\beta W - I} \rangle = e^{-\beta \Delta F}$$

Sagawa & Ueda, *PRL* **104**, 090602 (2010)

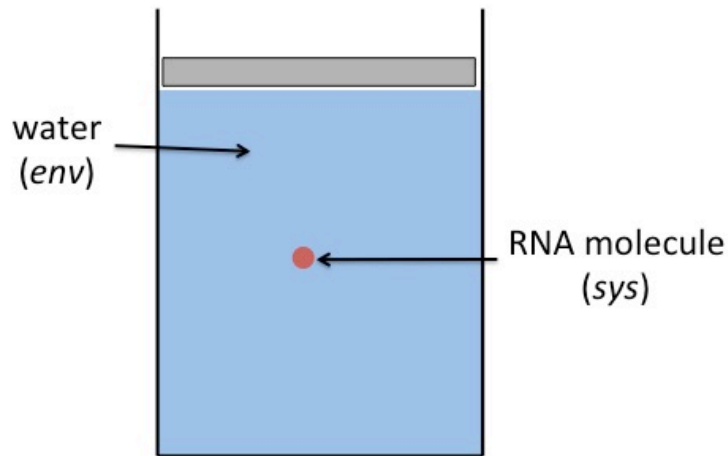
experiment:

Toyabe *et al*, *Nature Phys* **6**, 988 (2010)

Strong system-environment coupling

$$W \geq \Delta F \quad \left\langle e^{-\beta W} \right\rangle = e^{-\beta \Delta F} \quad \frac{\rho_F(+W)}{\rho_R(-W)} = \exp[\beta(W - \Delta F)]$$

- ΔF (Helmholtz) or ΔG (Gibbs) ? macro: $G = F + PV$
- How to define the volume of a single molecule ?
- How to define heat ? first law: $\Delta U = W - P\Delta V + Q$



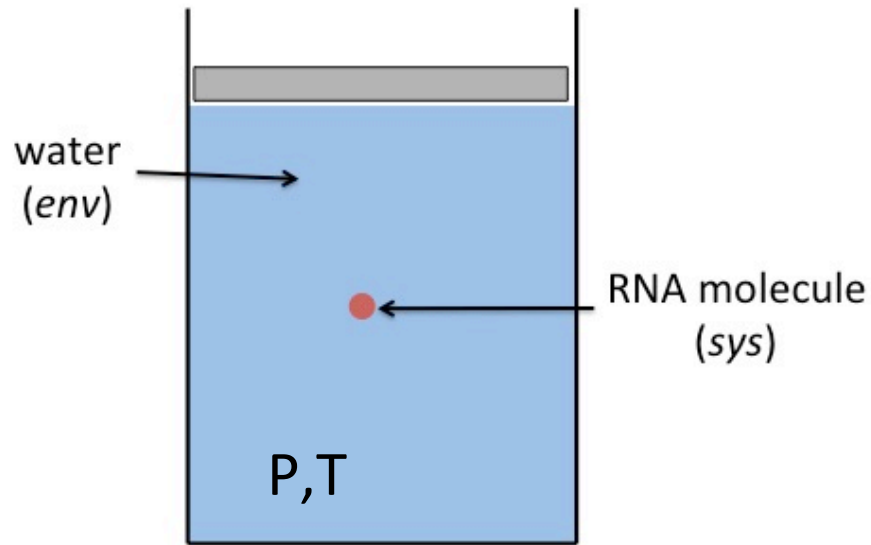
Total energy of sys + env:

$$U_{S+E} = U_{sys} + U_{env} + U_{int}$$

non-negligible!

Strong system-environment coupling

C.J., *PRX* 7, 011008 (2017)

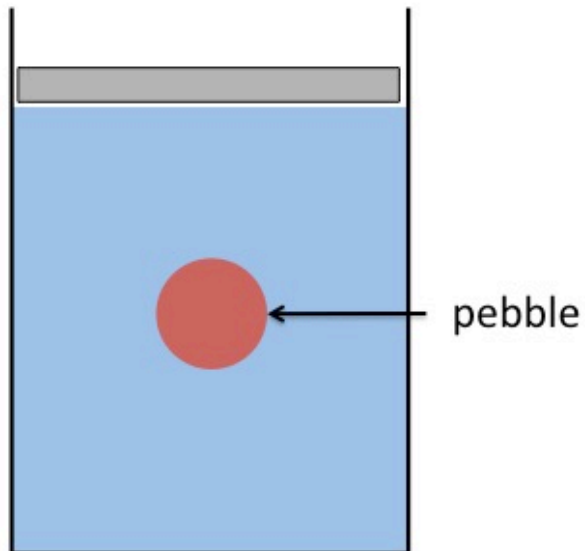


$$U_{S+E} = U_{sys} + U_{env} + U_{int}$$

$$p^{eq}(sys) = \frac{1}{Z} \exp[-\beta(U_{sys} + \phi)]$$

$\phi(q; P, T)$ = solvation potential
of mean force

↑
microscopic configuration of molecule



$\phi(q; P, T)$ = reversible work required
to insert pebble into water

$$= P \times V_{pebble}$$

$$V_{pebble} = \phi / P$$

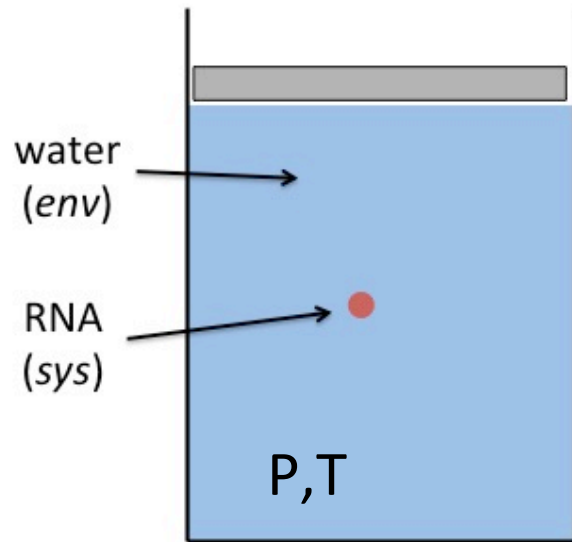
“thermodynamic
volume”

Strong system-environment coupling

C.J., *PRX* **7**, 011008 (2017)

Seifert, *PRL* **116**, 020601 (2016)

Strasberg & Esposito, *PRE* **95**, 062101 (2017)



define volume of system: $v(q; P, T) \equiv \phi / P$

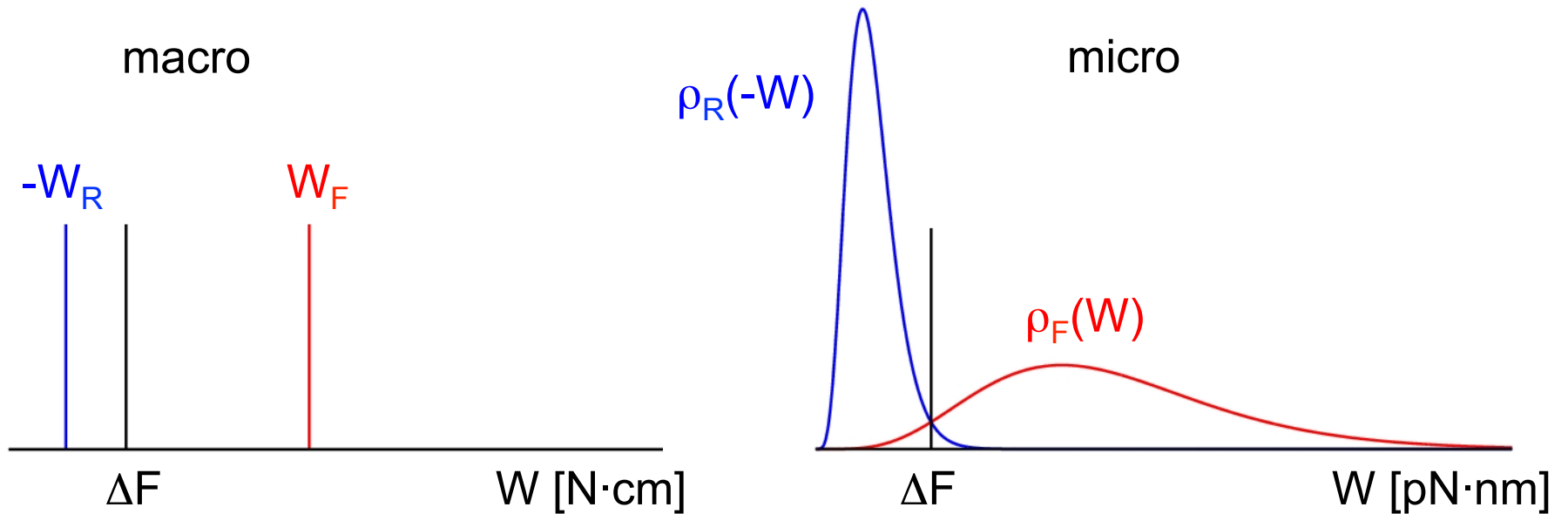
... leads to natural microscopic definitions of *internal energy, enthalpy, entropy, Helmholtz & Gibbs free energies, heat and work*

First law: $\Delta U_{sys} = Q + W - P\Delta v$

Second law: $\langle e^{-\beta W} \rangle = e^{-\beta \Delta G}$, $\frac{\rho_F(+W)}{\rho_R(-W)} = \exp[\beta(W - \Delta G)]$

$\langle W \rangle \geq \Delta G$, $\left\langle \int_A^B \frac{dQ}{T} \right\rangle \leq \Delta S$

Summary



$$W \geq \Delta F$$

$$-W_R \leq \Delta F \leq W_F$$

$$\langle e^{-\beta W} \rangle = e^{-\beta \Delta F}$$

$$\frac{\rho_F(+W)}{\rho_R(-W)} = \exp[\beta(W - \Delta F)]$$

... & others !

C.J., Annu Rev Cond Matt Phys **2**, 329 (2011) (*classical*)

Campisi, Hänggi, & Talkner, Rev Mod Phys **83**, 771 (2011) (*quantum*)

Sagawa, Progress Theor Phys **127**, 1 (2012) (*information processing*)